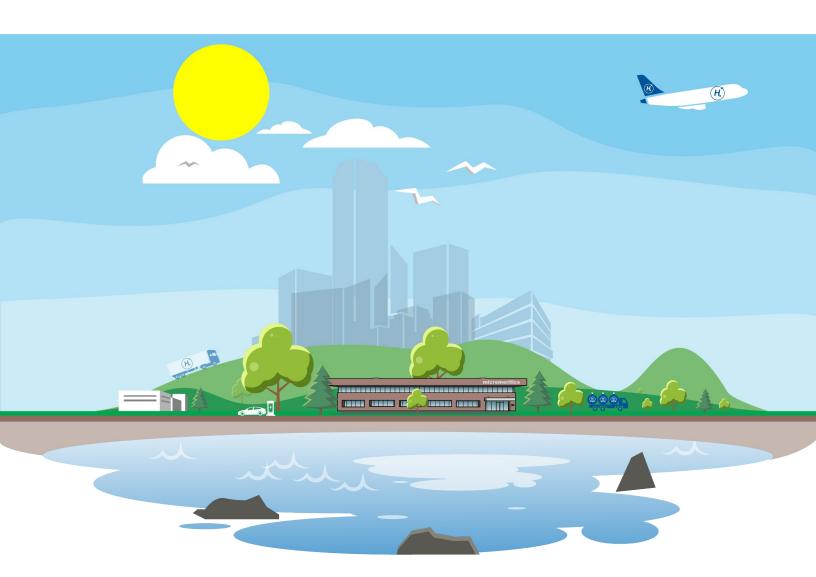
# NET-ZERØ TECHNOLOGIES

Micromeritics offers the <u>most comprehensive</u> portfolio of <u>high-performance instruments</u> to characterize the materials required to achieve a more <u>sustainable future</u>





# **HYDROGEN** LIFE CYCLE



Hydrogen will play a key role in decarbonization as it supports 60% of the applications with greenhouse gas (GHG) emissions.

Micromeritics products play a key role in the development of Adsorbents, Membranes, and Catalysts critical for technology development.

#### Adsorbents, Membranes, and Catalysts

- Optimize pore size of fuel cell membranes
- Use chemisorption to determine catalyst active area
- Adsorb/Desorb cycle optimization to minimize costs
- Study fuel cell efficiencies



# **HYDROGEN** PRODUCTION

**Steam Reforming** 



Blue Hydrogen is derived from natural gas with CO<sub>2</sub> capture and Green Hydrogen is produced by water electrolysis using renewable electricity.

#### Adsorbents, Membranes, and Catalysts

- Optimize adsorption / desorption cycle to increase productivity and reduce cost
- Determine CO<sub>2</sub> that can be adsorbed
- Maximize activity and lifetime of the catalyst
- Measure membrane pore size to optimize transport and reactivity

#### **Adsorbents, Catalysts**

- Develop materials with high H<sub>2</sub> adsorption
- Determine critical parameters to scale adsorbents
- Understand efficiency and lifetime of catalysts
- Maximize catalytic activity





Renewable and low carbon

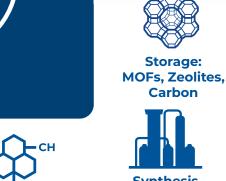
20% of global carbon

abatement by 2050

**Ammonia** Fertilizer, Fuel

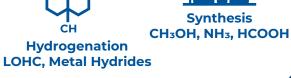
**Fuel Cells** 

Hydrogen to contribute over









**Green ElectroyIsis** 

**HYDROGEN** STORAGE



## **CARBON DIOXIDE** MITIGATION









Carbon capture, utilization, and storage, CCUS, is an important portfolio of emissions reduction technologies. A clean energy future includes electric vehicles, **valorizing CO<sub>2</sub>** for synthethic fuels, and industrial plants using carbon capture.

By **2050** almost **50%** of the **CO<sub>2</sub>** reductions come from technologies that are currently at the demostration or prototype phase.

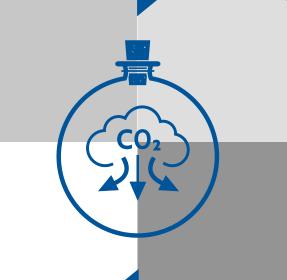


**Aviation E-kerosene** 

Shipping E-NH<sub>3</sub>, E-methanol

**Synthetic** 

**Fuels** 





**Direct Air Capture** 

**Industrial** 

Capture

**Amine Scrubber** 





#### **Adsorbents. Membranes**

- Effect of water on performance
- Tailor pore size of membrane for application
- Optimize adsorption / desorption cycle to minimize cost



- Evaluate effects of time-on-stream, temperature, and pressure on process economics
- Textural characterization of catalyst support
- Ascertain deactivation mechanisms
- Optimize metal dispersion and activity
- Determine reaction kinetics, activity, and selectivity of the catalyst





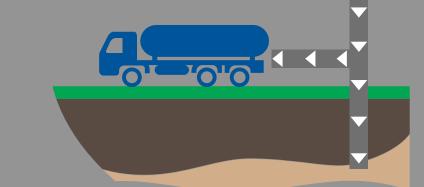


- Determine lifetime, cycling performance and adsorbent CO<sub>2</sub> capacity
- Understand local pollutants effect on adsorbent cycle life

 $CO_2$ UTILIZATION







### **ADSORBENT AND MEMBRANE SOLUTIONS**

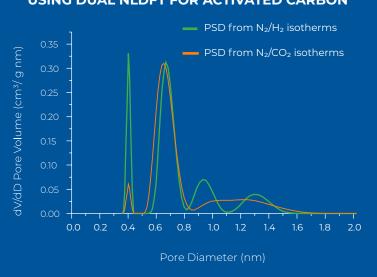
#### **3FLEX**

High-performance adsorption analyzer for measuring surface area, pore size and volume

- Understand adsorbent process cost using isoteric heat of adsorption
- Optimize pore size to maximize uptake capacity of the adsorbent
- Predict the selectivity of a gas mixture using Ideal Adsorption Solution Theory (IAST)



### COMPLETE PORE SIZE DISTRIBUTION (PSD) USING DUAL NLDFT FOR ACTIVATED CARBON



### **BreakThrough Analyzer (BTA)**

Precise characterization of adsorbents or membranes under process relevant conditions

- Lifetime and cycling studies to choose best adsorbent technology
- Measure kinetic performance of adsorbents
- Understand humidity effects for  $CO_2/N_2$  competitive adsorption

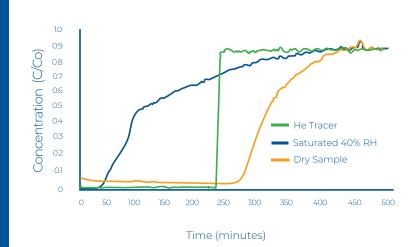
#### **AutoPore**

Mercury porosimetry analysis provides detailed porous material characterization

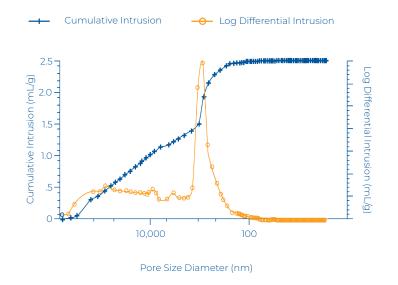
- Characterize pore size to understand diffusion into adsorption sights
- Study and optimize pore size distribution, total pore volume, percent porosity, particle size, and total surface area
- Assure reproducible adsorbent manufacturing process



#### CO<sub>2</sub> BREAKTHROUGH CURVES SIAI LOADED WITH PEI



#### NaY ZEOLITE CUMULATIVE INTRUSION VS PORE SIZE



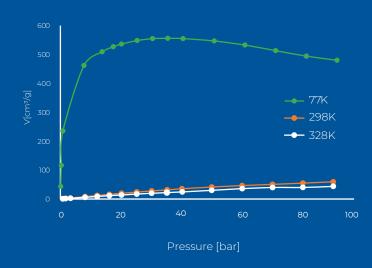
#### **HPVA\***

Static volumetric method to obtain high pressure adsorption and desorption isotherms

- Investigate the quantity of H<sub>2</sub> or CO<sub>2</sub> adsorbed
- Increase productivity and reduce cost by optimizing the adsorption / desorption cycle
- Study candidate materials and CO<sub>2</sub> storage sites



#### H<sub>2</sub> ADSORPTION ON MICROPOROUS CARBON



\* Not all products and configurations are available in all regions

### **CATALYST SOLUTIONS**

**AutoChem** 

materials' active sites

reaction conditions

**Utilizes dynamic techniques to characterize** 

• Optimize adsorption and dissociation

Understand if desorption occurs near

• Measure and quantify acid or base

sites to optimize reactivity and selectivity

of H<sub>2</sub>/O<sub>2</sub> on electrolysis electrodes

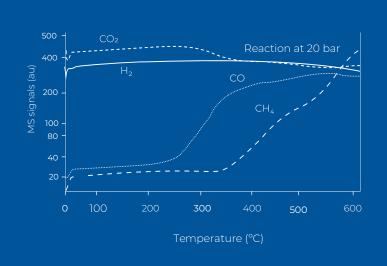
### FR/MR REACTOR SYSTEMS

### Benchtop reactor studies to understand and optimize catalyst performance

- Understand reaction kinetics to optimize operating parameters and conversion
- Measure selectivity, efficiency, and lifetime of catalysts
- Study reactions requiring gas / liquid separation at temperature and pressure



#### REDUCTION OF CO<sub>2</sub> IN THE SABATIER REACTION

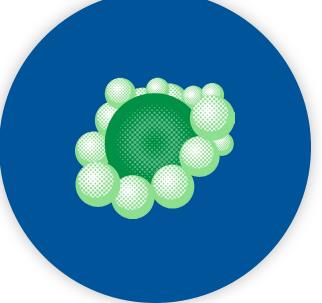


#### **ICCS**

### Provides in-situ characterization to understand the effect of reaction conditions on the catalyst

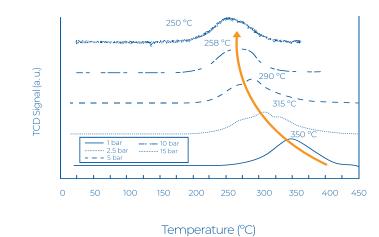
- Understand changes in performance over extended periods
- Determine deactivation mechanism to maximize the catalysts' lifetime
- Monitor changes in active sites, oxidative state, metal dispersion, and desorption behavior



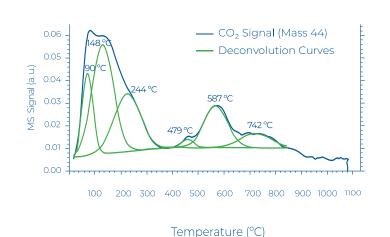




### PRESSURE IMPACT ON REDUCTION TEMPERATURE Cu-OXIDE CATALYST



#### DECONVOLUTION OF CO<sub>2</sub> DESORBED BY CaO/MgO



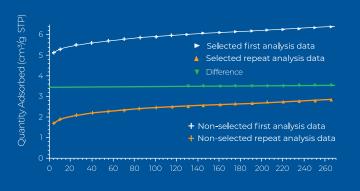
#### **3Flex** CHEMISORPTION

### Offers physisorption and static/dynamic chemisorption for characterizing catalysts

- Understand multi-metal catalysts' effects on activation and adsorption of active species
- Select catalysts providing a higher turnover frequency
- Investigate influence of heat of adsorption



#### ANALYSIS OF A SUPPORTED NI CATALYST USING H<sub>2</sub>



Pressure (mmHg)



# WORLDWIDE PRESENCE

micromeritics.com/worldwide

MORE than 12,000 installations in 100 different countries are used every day in the labs of the most innovative companies and the most prestigious government and academic institutions.

Customers choose Micromeritics for their world leading high-performance systems, expert application staff, and factory-trained engineers around the globe.

Get MORE from Micromeritics

# MATERIALS CHARACTERIZATION SERVICES WORLD-CLASS ACCREDITED LABORATORY





Need to characterize your materials or supplement your current lab's capabilities? Want access to top-of-the-line instruments and expert scientists?

The Micromeritics PTA lab is the leading contract laboratory for the characterization of adsorbents, catalysts, and membranes. The same engineers and scientists that develop and support our market-leading technologies are available to help you develop methods, test samples, and analyze the results.

- ISO 17025 accredited and FDA registered.
- Globally recognized scientists.
- Typical turnaround time: 7 business days
- Over 25 analytical techniques.

Contact PTA today to learn how our world-class laboratory can advance the development of your materials for the Net Zero economy.

Micromeritics products are 3rd party tested to conform to the highest level of compliance and safety. Visit micromeritics.com/compliance for full details by product.





### **Micromeritics Instrument Corporation**

4356 Communications Drive, Norcross, GA 30093 USA Tel.: +1 770-662-3636

© 2022 Micromeritics Instrument Corp. All rights reserved. All trademarks are the property of Micromeritics and its subsidiaries unless specified. The DNV logo is the property of Det Norske Veritas. The Intertek ETL logo is the property of Intertek. The IEC IECEE logo is the property of IEC. Specifications, terms and pricing are subject to change. Not all products are available in all countries. Please consult your local sales representative for details.